New Intramolecular Five-Endo-Mode Cyclization of Allenyl Aryl Ketones

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A convenient preparation of allenyl aryl ketones was achieved by the Weinreb-modified Grignard reaction of *N*-methoxy-*N*-methylamides with propargyl-magnesium bromide. On treatment with BF3•OEt2, the allenyl aryl ketones were converted to methylenebenzocyclopetenones *via* a new 5-endo-mode cyclization.

In the course of development of new SH-enzyme inhibitors, we found that treatment of diethyl (acetylamino)ethynylmalonates (1) with 1 M (M = molar conc.) KOH afforded trisubstituted oxazoles 3 via a new mode of 5-endo cyclization of the resultant acetylaminoallenic esters 2 under the basic conditions.¹⁾ We readily anticipated a possibility of a similar intramolecular 5-endo-mode cyclization²⁾ of allenyl aryl ketones under the Lewis acidic conditions for increasing electrophilicity of the conjugated allenic ketone moiety.

Now, we wish to report a convenient synthesis of conjugated allenyl aryl ketones utilizing the Weinreb-modified Grignard reaction³⁾ and their intramolecular 5-endo-mode cyclization. *N*-Methoxy-*N*-methylamides **4a-g**, prepared by a coventional method³⁾ with the corresponding carboxylic acids (or acid chlorides) and *N*,*O*-dimethylhydroxylamine hydrochloride, were treated with 1.5 mol equiv of propargylmagnesium bromide (1 M solution in Et₂O)⁴⁾ in anhydrous Et₂O at 0 °C and then room temperature with stirring to give the desired allenyl aryl ketones **5a-g** in high yields (70-88%) as shown in Table 1. Their allenic structure was confirmed by their characteristic ¹H NMR (200 MHz, CDCl₃) signals [δ 5.24-5.26 (d, 2H, J = 6.6 Hz) and 6.41-6.48 (t, 1H, J = 6.6 Hz)] due to allenic three protons. Although there have been some preparation methods for the conjugated allenyl ketones,⁵⁾ ours must be convenient.

Subsequently, intramolecular cyclization was attempted by employing an allenyl ketone **5a** and BF3•OEt2 as follows. To a solution of **5a** (741 mg, 3.6 mmol) in anhydrous CH2Cl2 (50 ml) was added BF3•OEt2 (447 µl, 3.6 mmol) at 0 °C with stirring under N2 atmosphere. After being stirred at 0 °C for 45 min, the reaction mixture was quenched with an aqueous NaHCO3 solution (40 ml). The mixture was treated as usual to give a bicyclic product **6a** (601 mg, 81%) as colorless prisms (Et2O-hexane = 1:1) after chromatographic purification.

Allenyl Aryl Ketone 5	Yield ^{a)} /%	mp/ °C	
5a	85	90-91	
5 b	88	70	
5 c	74	oil	
5 d	85	"	
5 e	70	40	
5 f	85	32-33 oil	
5 g	75		

a)All yields are those of isolated compounds.

$$\mathbf{a: R}^{1} = R^{2} = OMe, R^{3} = H$$

$$\mathbf{b: R}^{1} = R^{2} = OCH_{2}O, R^{3} = H$$

$$\mathbf{c: R}^{1} = R^{3} = OMe, R^{2} = H$$

$$\mathbf{d: R}^{1} = OMe, R^{2} = R^{3} = H$$

$$\mathbf{e: R}^{1} = R^{2} = Me, R^{3} = H$$

$$\mathbf{f: R}^{1} = Me, R^{2} = R^{3} = H$$

$$\mathbf{g: R}^{1} = R^{2} = R^{3} = H$$

Table 2. Intramolecular Cyclization of Allenyl Aryl Ketones 5a-g

Compd	Reaction co	Reaction conditions		Yield ^{a)} /%	mp / °C
5	Temp/°C	Time/min	6	1 1010 / 70	тр/ С
5a	0	10	6a	81 (p)	142-143
5 b	"	30	6 b	71 (<i>p</i>)	123-124
5 c	-138	20	6 c	50 (p)	89.5-90
5 d	0	"	6 d	$80 (p:o = 4:1)^{(b)}$	oil
5 e	60	40	6 e	$80 (p:o = 4:1)^{b)}$ $25 (p:o = 8:1)^{b)}$ $40 (p:o = 2:1)^{b)}$	"
5 f	"	45	6 f	$40 (p:o = 2:1)^{b}$	"
5 g	"	84	6 g	17 *	" "

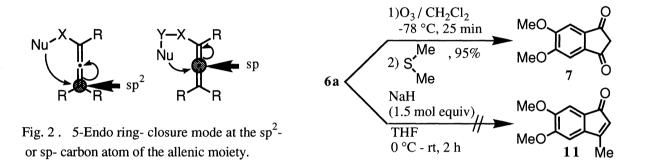
a) All yields are those of isolated compounds.

b) A ratio (para: ortho) was determined by the ¹H NMR (200 MHz, CDCl₃) analysis.

The reactions employing other compounds **5b-g** were similarly carried out to give the corresponding bicyclic products **6a-g** in various yields (Table 2). The chemical structure of compounds **6a-g** was determind on the basis of their ¹H NMR (200 MHz, CDCl₃) spectrum data and NOE expriments (400 MHz, CDCl₃) (*e.g.*, **6a** as shown in Fig. 1). Ozonolysis of **6a** followed by reduction of the resultant ozonide with dimethyl sulfide gave a 1,3-indandione derivative **7** [mp 226-227 °C (CH₂Cl₂-hexane); ¹H NMR (200 MHz, CDCl₃) δ 3.19 (s, 2H), 4.03 (s, 2H), and 7.34 (s, 2H)] in 95% yield. In this cyclization reaction, existence of an electron-donating substituent on the aromatic moiety seems to be essential. Steric interaction between the aromatic substituent and the allenic moiety may controll the regioselectivity (*para : ortho* ratio) in the cyclization. Five-endo ring-closure reactions (Nu = -0°) at the sp² carbon atom of the allenic moiety were independently reported by Macomber, ⁶ Magnus, ⁷) and Marshall⁴, ⁸) groups. Cyclization at the sp carbon atom has been briefly described by Torii group ¹) using SH nuclophile (Fig. 2). The intramolecular cyclization by our new method should proceed in

$$\begin{array}{c} \text{MeO} \\ \text{He} \\ \text{He} \\ \text{He} \\ \text{H}_{\text{d}} \\ \end{array} \begin{array}{c} \text{1H Signal} \\ \hline \delta \ 3.62 \ (\text{H}_{\text{e}}) \\ \delta \ 5.23 \ (\text{H}_{\text{d}}) \\ \delta \ 5.67 \ (\text{H}_{\text{c}}) \\ \delta \ 7.14 \ (\text{H}_{\text{b}}) \\ \hline \delta \ 7.18 \ (\text{H}_{\text{a}}) \\ \end{array}$$

Fig. 1. NOE Experiment (400 MHz, CDCl₃) and ¹H NMR (200 MHz, CDCl₃) of compound **6a**.



$$\begin{array}{c} \text{MeO} \\ \text{MeO} \\ \text{3.4\%} \\ \text{10} \\ \end{array} \begin{array}{c} \text{BF}_3 \bullet \text{OEt}_2 \\ \text{(1 mol equiv)} \\ \text{MeO} \\ \end{array} \begin{array}{c} \text{MeO} \\ \text{MeO} \\ \text{MeO} \\ \end{array} \begin{array}{c} \text{MeO} \\ \text{MeO} \\ \text{MeO} \\ \end{array} \begin{array}{c} \text{R} \\ \text{R} \\ \text{R} \\ \end{array} \begin{array}{c} \text{CH}_2\text{Cl}_2 \\ \text{reflux}, 5 \text{ h} \\ \text{TiCl}_4 \\ \text{(1 mol equiv)} \\ \text{CH}_2\text{Cl}_2 \\ \text{reflux}, 5 \text{ h} \\ \end{array} \begin{array}{c} \text{79\% recovery} \\ \text{of 8} \\ \end{array} \\ \text{R} \\ \end{array} \begin{array}{c} \text{R} \\ \text{TiCl}_4 \\ \text{(1 mol equiv)} \\ \text{CH}_2\text{Cl}_2 \\ \text{of 8} \\ \end{array} \begin{array}{c} \text{73\% recovery} \\ \text{of 8} \\ \end{array}$$

the 5-endo-mode at the sp carbon atom of the allenic moiety because all reactions furnished exo-methylene products. Although there have been many papers on the Nazarov type reaction, 9) cyclization of alkenyl aryl ketones such as compounds 8 and 9 has never been reported. 10) In order to compare with our new 5-endo-mode cyclization, 5-endo-trigonal like ones were attempted by employing 8 and 9 in the presence of BF3•OEt2 or TiCl4. However, these reactions resulted in 73% or 79% recovery of the starting compound and production of a small amount of 10 with miscellaneous compounds. Interestingly, the exo-methylene double bond of 6a-g must be remarkably stable when compared with that of the corresponding conjugated enones like 11. Isomerization of the exo-methylene double bond of 6a toward 11 does not occur at all under the basic conditions. This unusual stability of 6a-g is quite different from the case of simple methylenecyclopentanone reported by Allinger et al. 11)

Thus, we demonstrated a new 5-endo-mode cyclization of allenyl aryl ketones $^{12)}$ which will be expanded to preparation of larger cyclic compounds. Compounds **6a-d** may be available for synthetic development of new antitumor agents similar to the duocarmycin analogs. $^{13)}$

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